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Synthesis of 1-β-L-Arabinofuranosylcytosine, the Enantiomer of Cytosine Arabinoside¹

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Wu and Chargaff,² in a study of the enantiomers of natural nucleosides, have determined that L-uridine is an acceptor for phosphate transfer from carrot phosphotransferase as well as nucleoside phosphotransferase from human prostate. In a similar study using Ladenosine, Shimizu and coworkers³ have determined that L-adenosine was 50% deaminated in the time that the *D* enantiomer was completely converted to D-inosine. Snake venom 5'-nucleotidase did not accept L-adenylic acid as a substrate. It is apparent from these examples that enzymes differ widely in their ability to accept L nucleosides as substrates. It was therefore of interest to prepare the enantiomer of the anticancer agent, cytosine arabinoside, since the D enantiomer is known to undergo a rapid enzymatic deamination to form an inactive metabolite. $1-\beta$ -Darabinofuranosvluracil.4

An extension of the procedure of Sanchez and Orgel⁵ has provided a novel approach to the synthesis of the β -L-nucleoside. Treatment of L-arabinose with cyanamide in MeOH containing coned NH₄OH gave 2-amino- β -L-arabinofurano [1',2'-4,5]-2-oxazoline (1) in good yield. Ring closure of 1 with cyanoacetylene furnished a cyclonucleoside intermediate, which was hydrolyzed, by NH₄OH without isolation, to 1- β -Larabinofuranosyl cytosine hydrochloride (2).



Anticancer Evaluation.—Results received to date indicate that 2 possesses no significant activity⁶ against lymphoid leukemia L1210 or the Ridgway osteogenic sarcoma. **Enzymatic** Investigation.—1- β -L-Arabinofuranosylcytosine \cdot HCl did not function as a substrate in an *Escherichia coli* cytidine aminohydrolase system.⁷

Experimental Section⁸

2-Amino- β -L-arabinofurano[1',2'-4,5]-2-oxazoline (1).—To a suspension of L-arabinose (30.0 g, 0.2 mole) in MeOH (100 ml) was added cyanamide (16.8 g, 0.4 mole) and concd NH₄OH (10 ml). The stoppered flask has stirred at ambient temp for 24 hr and cooled to 5°, and the solid was filtered and washed with cold *i*-PrOH (24.3 g, 70%). The anal. sample was recrystd from aq MeOH: mp 175° (dec with bubbling), $[\alpha]^{25}D + 16.1°$ (c 1.0, H₂O); no uv spectrum above 220 nm. Anal. (C₆H₁₀N₂O₄) C, H, N.

1-β-L-Arabinofuranosylcytosine (2).—A suspension of 1 (6.96 g, 0.04 mole) in dimethylacetamide (20 ml) was cooled in an ice bath and cyanoacetylene⁹ (2.5 ml, 0.04 mole) was added by syringe through a serum cap to the partially evacuated flask. The reaction mixt was allowed to warm to room temp and after 40 min was poured into 1 N NH₄OH (100 ml). The soln was heat at 70° for 15 min and the dark mixt was then evapd to dryness *in vacuo*. Two 25-ml portions of MeOH were successively added and evapd *in vacuo*. Addition of 3% dry HCl in MeOH and vol reduction caused crystn. Filtration furnished a crude solid which was recrystd with charcoal from MeOH-EtOAc (7.26 g, 68%). The anal. sample was recrystd from aq *i*-PrOH: mp 197° dec, [α]²⁵D - 127.7° (c 1.0, H₂O); λ^{mi1}_{max} 280 nm (ε 14,000), λ^{mi7}_{max} 272 nm (10,700). Anal. (C₃H₁₂N₃O₅·HCl) C, H, N.

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(8) Satisfactory analytical data (C. H. N within $\pm 0.4\%$ of theoretical values) were obtained from MHW Laboratories, Garden City, Mich. Melting points were determined on a Thomas-Hoover melting point apparatus and are uncor. The uv spectra were recorded on a Cary 15 spectrophotometer and optical rotations were obtained with a Perkin-Elmer Model 141 automatic digital readout polarimeter.

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Preliminary Studies on the Antitumor Activity of Some Phosphatidyl Nitrogen Mustard Derivatives

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As part of our studies on exploring the use of phospholipid moieties for the potential dual function of transport and latentiation of the bis(2-chloroethyl)amino group, we synthesized a number of phosphatidyl nitrogen mustard intermediates (**3a-3h**, Table I) following the general procedure of Batrakov, *et al.*,¹ who have synthesized **3c** but did not report on its antitumor activity.¹ Derivatives **3a-3g**, as well as nonlipid synthetic precursors **1** and **2** were tested against leukemia L-1210 in mice² and found to afford

⁽¹⁾ Synonyms for cytosine arabinoside are: Ara-C, cytarabine, and 1- β -D-arabinofuranosylcytosine.

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⁽²⁾ Screening data were furnished by Vitro Laboratories. Silver Spring, Md., under contract to the Cancer Chemotherapy National Service Center; 8h was not available in sufficient quantity for screening.

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			CHOCO(C	$H_2)_n CH_3$			
			$\operatorname{CH}_{2}\operatorname{OP}(O)$	$N(CH_2CH_2Cl)_2$			
			OC H	. ,-			
			00611	5			
			Yield,				
\mathbf{Compd}	n^a	Empirical formula	%	Mp, °C	$n^{20}D$	R_{f}^{b}	Analyses ^c
3a	16	$C_{49}H_{88}Cl_2NO_7P$	58	29-32		0.33	C, H, N
3b		$C_{49}H_{84}Cl_2NO_7P$	30		1.4807	0.30	C, H, N
3c	14	$C_{45}N_{80}Cl_2NO_7P$	53	37-39		0.33	C, H, N
3d	12	$C_{41}H_{72}Cl_2NO_7P$	60	39-40		0.30	C, H, N
3e	10	$\mathrm{C_{37}H_{64}Cl_2NO_7P}$	59		1.4802	0.30	С, Н
3f	8	$\mathrm{C}_{33}\mathrm{H}_{56}\mathrm{Cl}_2\mathrm{NO}_7\mathrm{P}$	59		1.4803	0.27	C, H, N
3 g	7	$C_{31}H_{52}Cl_2NO_7P$	54		1.4838	0.27	C, H, N
3h	6	$C_{29}H_{48}Cl_2NO_7P$	25		1.4800	0.26	C, H, N
01			1.1 (0, 1.1	(1, 1) 1	11 · D1 ·		T 0 (00)

TABLE I 1,2-Diglycerides 3-[Hydrogen bis(2-chloroethyl)phosphoramidate phenyl ester] CH₂OCO(CH₂)₂CH₂

 a 3b = 1,2-disubstituted unsatd olein [hydrogen bis(2-chloroethyl)phosphoramidate Ph ester]. Silica,⁵ hexane-Et₂O (60:40). ^c See ref 5.

no significant prolongation of survival time. These results strongly indicate that lack of activity is this tumor of the "cytotoxic" N mustard moiety of these analogs is not related to significant differences in lipid solubilities or other physicochemical properties, and broadly confirm literature reports^{8,4} that antineoplastic activity in phosphoramidate and phosphoroesteramidic mustards is greatly diminished or abolished by substituents such as Ph.

Experimental Section⁵

N,N-Bis(2-chloroethyl)phosphoramidic acid (2,2-dimethyl-1,3dioxolanyl-4-yl)methyl phenyl ester (1) and N,N-bis(2-chloroethyl)phosphoramidic acid 2,3-dihydroxypropyl phenyl ester (2) were prepd by the method of Batrakov, *et al.*,¹ upon prepn 2 was used directly for acylation reactions. Deriv **3b** was prepd according to the procedure described for the synthesis of **3a-3h** except that acylation was carried out at 70°.¹ Spectral data (ir, pmr) were nearly identical for each **3** homolog and are reported only for **3c**. Some physical properties of **3a-3h** are included in Table I.

1,2-Diglycerides 3-[Hydrogen bis(2-chloroethyl)phosphoramidate, phenyl ester] (3a-3h).—To 1.48 g (0.004 mole) of 2 in 8 ml of pyridine (0°) was added 3.3 g (0.012 mole) of palmitoyl chloride, and the mixt was kept stirring at room temp for 48 hr. The reaction mixt then was poured into ice H₂O and the solid material was extd with three 25-ml portions of Et₂O. The exts were combined and washed successively with H₂O, ice-cold 2% H₂SO₄, and again with H₂O. The soln was dried (MgSO₄), filtered, and concd under reduced pressure to give 4 g of a crude yellow oil, which was placed on an Al₂O₈ (100 g) column (60 × 2 cm) and was eluted with 600 ml of CHCl₃. Concn of the eluate gave 2.6 g of a yellow viscous liq which showed a major (R₁ 0.3) and a minor (R₁ 0.6) spot on tlc (silica gel, hexane-Et₂O, 60:40). The viscous material then was chromatogd on a silica gel (50 g) column (60 × 2 cm) with hexane-Et₂O (60:40) as eluent (450

(5) Fatty acyl chlorides were obtained from Eastman Kodak Co., Rochester, N. Y. 14650, and Sigma Chemical Co., St. Louis, Mo. 63118, and were used without further purification. Silicar tle 7GF (Mallinckrodt) was used for tle analyses; column chromatog purifications were made with aluminum oxide, neutral, and silica gel (E. Merck AG). The ir spectra (neat, Nujol) were detd with Beckman IR-8 and Perkin-Elmer 521 spectrophotometers, and pur spectra with Varian A-60 and HA-100 spectrometers using CDCls (Me₂Si) as solvent. Melting points were detd with a Thomas-Hoover capillary melting point apparatus and are uncor. Refractive indices were detd with a Bausch and Lomb Abbe-3L refractometer. Where analyses (Table I) are indicated only by the symbols of the elements, anal. results obtained for those elements were within $\pm 0.4\%$ of the theor values. Analyses were performed by Micro-Analysis, Inc., Marshallton, Wilmington, Del. ml). Fractions which showed one spot (R_f 0.33) were combined and concd under reduced pressure to obtain 1.8 g (53%) of a colorless oil. The oil was dissolved in 15 ml of petr ether (bp 30-80°) and kept at -10° for 8 hr, during which time an amorphous, white solid sepd. Recrystn using Norite A (boiling, 5 min) gave **3c**: mp 37-39° (lit.¹ 35.5-37°); R_f 0.33, silica gel, hexane-ether (60:40); ir (Nujol) 1748 (C=O), 1262 (P=O), 1200 (POC_{arom}) 1090 (PN), 1060 (POC_{aliph}), 762 cm⁻¹ (CCl); 100-MHz pmr (CDCl₃) τ 2.64-2.85 (m, 5), 4.73 (t, 1), 5.52-5.98 (m, 4) 6.33-6.88 (m, 8), 7.72 (t, 4), 8.75 (s, 52) 9.14 (t, 6).

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New Thio Derivatives of Carcinogenic Arylamines. 5. Ring-Substituted Methylthio-4-acetamidostilbenes^{1a}

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In earlier papers in this series we described the synthesis of some new thiofluorenes^{1b-d} related to the metabolism of the carcinogen 2-acetamidofluorene, and of a similar new thio derivative of 4-acetamidodiphenyl.^{1e}

The carcinogen N-hydroxy-4-acetamidostilbene [N-(OH)-4-AAS] yielded a more complex metabolic pic-

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